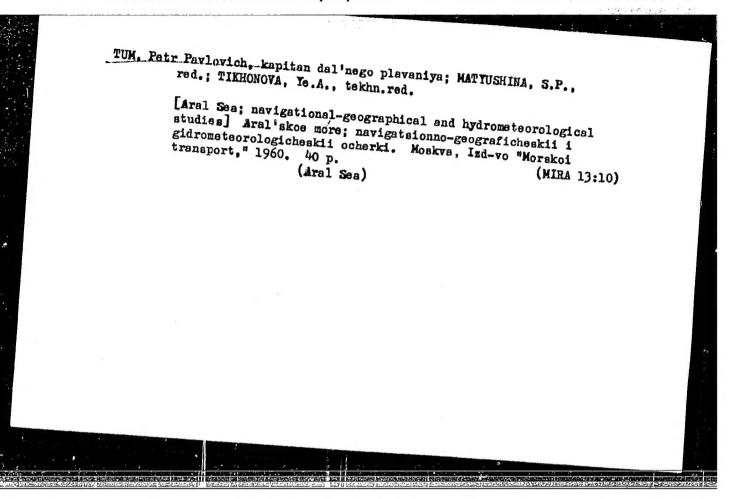
- 1. TUL'ZHENKOVA, F. F.
- 2. USSR (600)
- 4. Vegetable Gardening Arctic Regions
- 7. Vegetable gardening in the Arctic. Sad i og. no. 11, 1952

9. Monthly List of Russian Accessions, Library of Congress, March 1953, Unclassified.

TULIZIANEOTA, T. F.

2/532 Byrashchivaniye ogurtsov i lika vo mkhu. Sad i ogorol, 1949, No. 8, c. 65-67

50: LETOFIS' NO. 35, 1949



APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001757420007-8"

TUM, P.

Poultry - Feeding and Feeding Stuffs

Salt in increasing productivity of ducks and chickens. Sots. thiv. 14 no. 8, 1952.

Monthly List of Russian Accessions, Library of Congress, November 1952. UNCLASSIFIED.

TUMA, A.

Auto accessories at the Paris automobile axhibition

P. 175 (Motoristicka Soucasnost) Vol. 3, No. 2, May 1957, Czechoslovakia

SO: MONTHLY INDEX OF EAST EUROPEAN ACCESSIONS (EEAI) LC. - VOL. 7, NOL 1, JAN. 1958

Autumn congresses of international of motorists organizations. p. 713.

(SVET MOTORE, Vol. 10, No. 23, Nov 1956, Praha, Czechoslovakia)

SO: Monthly List of East European Accessions (EEAL) LC, Vol. 6, No. 12, Dec 1957. Uncl.

TUMA . A

Workshops of the League for Cooperation with the Army.

P. 21. (SVET MOTORU.) (Praha, Czechoslovakia) Vol. 12, No. 1, Jan. 1958

SO: Monthly Index of East European Accession (EEAI) LC. Vol. 7, No. 5, 1958

Tuma, A.

"Tragedy on the highways." p.379

SVET MOTORU. (Svaz pro spolupraci s armadou) Praha, Czechoslovakia, Vol. 13, no. 12, June 1959.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, No. 9, Sept. 1959 Uncl.

TUMA, A.

"In the streets of Prague."

p. 58 (Svet Motoru) Vol. 12, no. 2, Jan. 1958
Prague, Czechoslovakia

S0: Monthly Index of East European Accessions (EEAI) IC. Vol. 7, no. 4,
April 1958

17. 法计算器的图

TUMA, A.

Shall we have reflective traffic signs? p. 219.

SVET MOTORU. (Svaz pro spolupraci s armadou) Praha, Czechoslovakia Vol. 13, no. 7, Mar. 1959

Monthly list of East European Acessions (EEAI), LC, Vol. 8, no. 7 July 1959 Uncl.

TUMA, A.

Cur first congress. p. 161. (Svet Motoru. Praha. Vol 10, no. 6, Mar. 1956.)

SO: Monthly List of East European Accessions (EEAL) LC., Vol. 6, no. 7, July 1957. Uncl.

Tuma, A.

Tuma, A. For development of motorism. p. 257.

Vol. 10, no. 9, Apr. 1956 SVET MOTORU TECHNOLOGY Czechoslovakia

So: East European Accessions, Vol. 6, May 1957

THEA, A.

Motor sport for everybody. p. 321. UVE: MOTORY, Praha, Vol. 9, no. 11, kay 1955.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955, Uncl.

TUMA, A.

Preparations for the 30th Annual International Six-Day Motorcycle cace, September 13-18. p. 362.

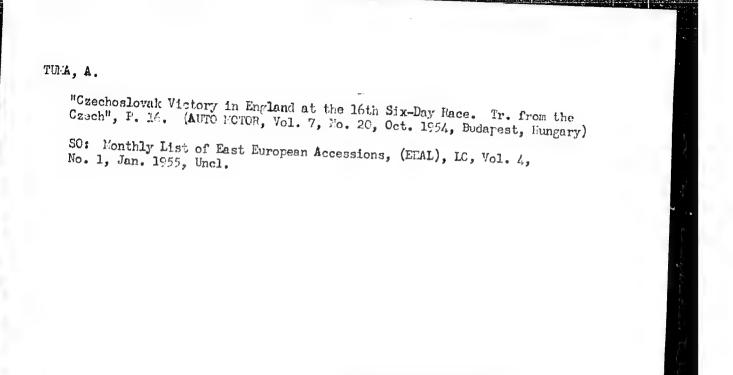
SVET MOTORU, Praha, Vol. 9, no. 12, June 1955.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955, Uncl.

TUMA, A.

Let us prepare well for the Six-Day Race. p. 417. Strikes in American automobile factories. Tr. from the Russian. p. 418. Catastrophe in Le Mans. p. 418. SVET MOTORU, Praha, Vol. 9, no. 14, July 1955.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955, Uncl.



TUMA, A.

"The Africa of Hanzelka and Tuma." p. 30 (Svet Motoru, Vol. 7, no. 137, Jan. 1953, Praha)

SO: Monthly List of East European Accessions, Vol. 3, no. 2, Library of Congress, Feb. 1954, Uncl.

SRACKOVA, D.; TEYSCHL, O.; TUMA, A.

Value of the determination of bilirubin fractions in icterus neonatorum. Cesk. pediat. 20 no.10:849-854 0 '65.

l. II. detska klinika lekarske fakulty University J.E. Purkyne v Brne (prednosta prof. dr. M. Toman) a Ustredni laborator Fakultni detske nemocnice v Brne (vedouci MUDr. O. Teyschl).

TUMA, ADOLF.

Tuma, Adolf, Od kocaru k automobilu; 50 let automobilky Tatra v Koprivinici. (V Praze, Tatra, 1947) lv. (unpaged) (From the carriage to the automobile; 50 yrs. of the Tatra Automobile Plant in Koprivince. English and French, German, and Russian summaries. illus.)

SO: Monthly List Of East European Accessions, LC. Vol.3, No. 5 May 1954, Unclassified

TUMA, B.

Let us attain higher efficiency in handling heavy building machinery.

P. 189 (Mechanisace) Vol 4, No. 6, June, 1957, Czechoslovakia

SO: MONTHLY INDEX OF EAST EUROPEAN ACCESSIONS (EEAI) LC. - VOL. 7, NO. 1, JAN. 1958

TUMA, B.

Cholevik. Deficiencies of dynamo starters of the DKW and JAWA vehicles. p. 205. SVET MOTORU, Praha, Vol. 9, no. 7, Mar. 1955.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955, Uncl.

TOMSA, Antonin, inz.; TUMA, C.O., inz.

General amplifier for measuring and recording of vibrations. Letecky obzor 6 no.3:90-91 \*62.

Z/040/62/000/003/003/003 D006/D102

AUTHORS:

Tomsa, Ant., Engineer, and Tuma, C. O., Engineer

TITLE:

Universal amplifier for vibration measuring and recording

PERIODICAL: Letecky obzor, no. 3, 1962, 90

TEXT: The article presents a description of a universal amplifier with an input resistance of 2 M  $\Omega$  and four outputs for a pointer meter, a recorder, a cathode-ray oscilloscope and a loop oscillograph, respectively. It can be used for measuring vibrations of engines and machines, for dynamic stress measurements using strain gauges, or as a high-sensitivity amplifier for various other low, alternating voltages. The amplifier sensitivity is such that vibration amplitudes as low as low can be determined and registered. The instrument is a low-frequency, three-stage amplifier with a filter suppressing frequencies below 50 cps, four cathode followers, and a built-in pointer meter. Its total gain is  $10^3$ . The input sensitivity can be varied from 5 to 500 mV in seven steps. The instrument has a built-in voltage stabilizer permitting its plugging into a 220 V, 50 cps

Card 1/2

Z/040/62/000/003/003/003 D006/D102

Universal amplifier ...

line with ± 10% deviation. Some of these amplifiers already are in operation at the ZJŠ in Prague-Jinonice, and the Rukov - sdružený podnik okresu Děčín, závod 122 Varnsdorf (Rukov - Associated Enterprise of the Děčín District, plant 122, Varnsdorf) will produce a larger number of these amplifiers. There are 4 figures.

Card 2/2

TUMA F. VAVRIN, F.

Quick-hardening cement. p. 76.

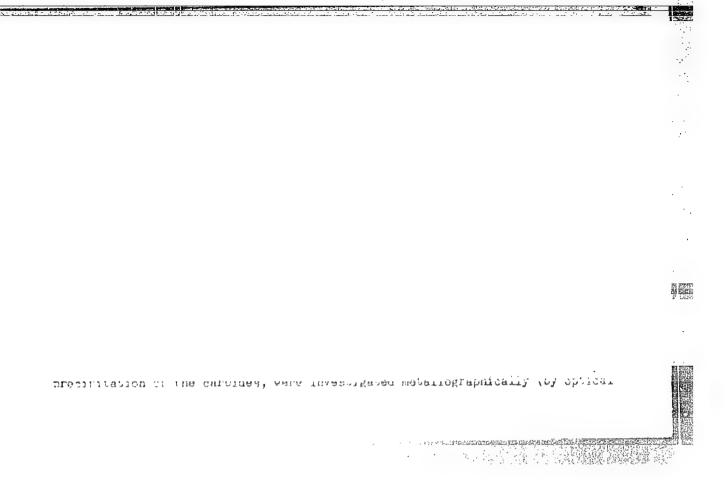
STAVIVO. (Ministerstvo stavebnictvi) Praha, Czechoslovakia, Vol. 37, no. 3, Mar. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, no. 7, July 1959 uncla.

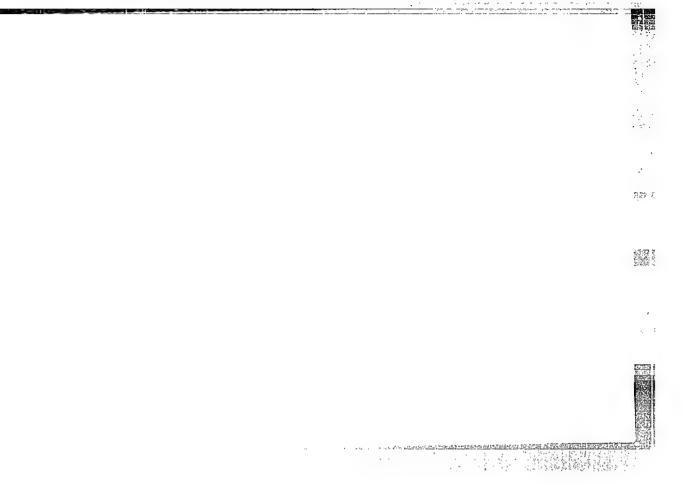
Hortars and grouts for panel assembling under low temperature.

Foz stavby 11 no.5:264-267 \*63.

1. Vyzkumny ustav stavebni, Gottwaldov.



20





VYKLITSKIY, M.; KRALIK, F.; TUMA, G.

Distribution of elements in the & - and /-phases of chromium nickel austenitic and ferritic steels. Avtox. evar. 17 no.2:30-37 F '64. (MIRA 17:9)

1. Gosudarstvenmyy nauchno-issledovatel'skiy institut materialov i tekhnologii Akademii nauk Chekhoslovatskoy Sotsialisticheskoy Respubliki.

2/0065/64/000/001/0013/0027

AUTHOR: Vyklicky, Miloslav (Vy\*klitskiy, Miloslav); Kralik, Frantisek (Kralik,

Frantishek); Tima, Hanus (Tima, Ganush)

TITLE: Distribution of the elements between the alpha and gamma phases in chromium-nickel steels with two-phase structure

SOURCE: Kovove materialy, no. 1, 1964, 13-27 TOPIC TAGS: element distribution, alpha phase, gamma phase, chromium-nickel steel,

ABSTRACT: The paper studies with a KAMEKA micro-probe the distribution of manganese and and nickel in ferrite and austenite in two-phase chromium-nickel steels chromium and nickel in territe and austonites in two-phase different steels with a content of about 0.1% C. 21% Cr. 0.5--9.8% Mn, 3.1--6.6% Ni, some of which with a content of about 0.1% 6, 21% 67, 0.3-y.0% MI, J.1-0.0% MI, some of whiteh were further alloyed with about 2% Mo and 0.3% Ti. It was found that the distribuwere runner arroyed with about to no and 0.77 it. It was found that the distribution factor in the range of chemical composition studied is approximately constant; tion factor in the range of chemical composition studied is approximately constant about 1.2 for chromium, and 0.9 for manganese. For nickel, this factor depends about 1.2 for chromium, and 0.9 for manganese. For nickel, this factor depends about 1.2 for chromium, and 0.9 for manganese. The heat of solution was found to be about +500 cal/mol for chromium, about -300

for manganese and from -1,000 to -1,500 for nickel, depending on the nickel content. The data determined for chromium and nickel agreed well with those cited in the literature. The value of -2,040 cal/mol given for manganese in the literature is based on balanced binary Fe-Mn diagrams, where the breakdown of the manganese into alpha and gamma phases is determined indirectly (dilatometrically, metallographically, etc.), and conflicts with all practical experience thus far gained. The paper also shows that in the alloys studied the heat of solution depends on the temperature, which contradicts Zener (Transactions of the Am. Inst. of Mining and Metall. Engineers, 167, 1946) and Jones and Pumphrey (J. Iron and Steel Inst., 163, its distribution does not depend either on the temperature or on the concentration of the alloy elements. The authors could not decide from their experiments whether specimens or whether that assumption was unjustified. Original has 6 tables, 8

ASSOCIATION: Statni vyskumny ustav materialu a technologie, Prague (State Experimental Establishment for Material and Technology); Laboratorium fyziky kovov SAV, Bratislava (Laboratory for the Physics of Metale of the SAV)

Card 2/87

Z/0032/64/014/007/0509/0517

AUTHOR: Vyklicky, M. (Engineer); Mericka, M., Kabrhel, A. (Engineer); Tuma, H., (Engineer); Kopal, V. (Engineer); Mursec, M. (Engineer); Dvorak, K. (Engineer); Vältr, V.

TITLE: Corrolion resistance of steel with a two-phase structure of the type Cr21Ni5

SOURCE: Strojirenstvi, v. 14, no. 7, 1964, 509-517

TOPIC TAGS: chromium steel, nickel steel, stainless steel, corrosion resistance, phase structure, alloy steel, alloying, phosphorus, titanium

ABSTRACT: Extensive experiments have been carried out to test corrosion resistance of newly introduced non-rusting steels with a two-phase structure of the type Cr2lNi5, which are mainly utilized in equipment of the chemical industry. The tests were carried out in the laboratory and confirmed by experiments in industrial plants, and included comparisons with classical steels which the new types

Card 1/5

were to replace. Laboratory tests of the usual type were carried out on 30 x 80 x 2 mm (and also 1 mm) samples and plant tests on 20 x x 100 x 2 mm samples. The results of the experiments are in agreement with corrosion theory. Increased phosphorus content lowers the corrosion resistance. The varying effect of titanium added to Cr21Ni5 and Cr18Ni9 in different acids is discussed. In general it is found that the optimal types of two-phase steels have a corrosion resistance similar to that of classical austenite steel while being more economical than the corrosion resistant CSN 17460 and 17471 steels, and exhibiting a much higher intercrystalline-corrosion resistance. It was found that in the food-processing industry Cr21Ni5Ti can almost fully replace CSN 17246 steel. Orig. art. has: 6 figures and 13 tables.

ASSOCIATION: SVUMT, Prague

SUBMITTED: 00

ENCL: 03

SUB CODE: MM

NR REF SOV: 001

OTHER: 006

Card 2/5

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Oool 1 '-	ČSN 17784 Cržiniati		CBN 42 2953 CBN 42 2928 Cr21N15 Cr21N15Ti		CBN 42 2013 Cr21NIAMO2	
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p/mm <sup>4</sup> ] min.	40	38	38	35	35	36
p/mm³]	6590	6590	6590	68—90	65—90	65—90
[%] min.	23	20	20	18	12	
%] min.	35	35	35	15 .	, 10	18
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rdost HB 7 7	<u>-</u>	-	-	180—250	180-250	180—250

ENCLOSURE: 02

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Topolné zpracování*) 10	980—1020 °C	-1020 °C 1000—1050 °C		980—1020 °C	
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Card 4/5

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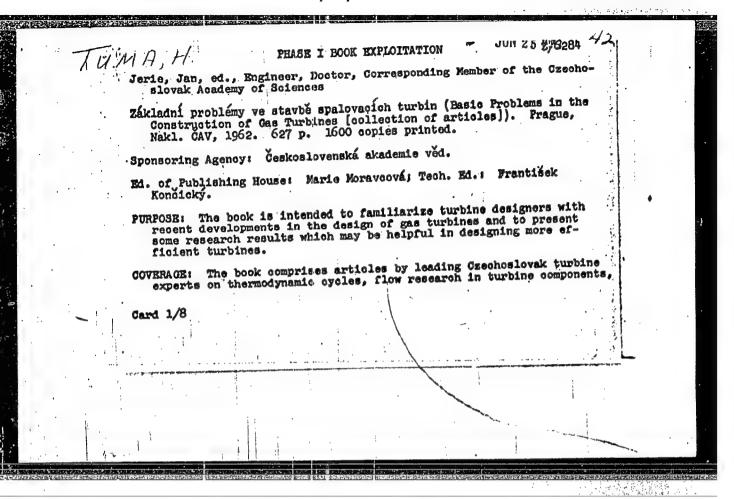
ENCLOSURE: 03

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Legend for Enc. 01: 1 - steel, 2 - article, 3 - plate, 4 - rod, 5 - forging, 6 - casging, 7 - hardness, 8 - weldability, 9 - electrodes used, 10 - heat treatment, 11 - maximum operating temperature, 12 - replaces CSN steel, 13 - workability, 14 - guaranteed, 15 - good

"With suggested quenching in water or air. Steels with two-phase structure are more workable than conventional pure austenitic steels.

Card 5/5



		Basic Problems in the Construction (Cont.) 2/6	284	3	
	:	J. Voscdalek (State Research Institute for Materials and Technology, Prague). Requirements for Construction Materials of the Principal Turbine Components	183		
		L. Čížek and M. Vystyd (State Research Institute for Materials and Technology, Prague). Current State and Development of Heat-Resistant Materials for Gas Turbines	199		•
		L. Cizek. Prospective Materials for Use in Gas Turbine Con- struction	211		÷
,	1.	Z. Eminger (V. I. Lenin Plant, Plzen) and J. Krumpos (State Research Institute for Materials and Technology, Prague).  The Austenitic Alloy "IZ"	221		
	1		4.0	i	•
		M. Vystyd, J. Jožek, and H. Tuma (State Research Institute for Materials and Technology, Prague). The Relationship between the Microstructure and the Properties of Some Heat-Resistant Steels and Alloys	233		
		Materials and Technology, Prague). The Relationship between the Microstructure and the Properties of Some Heat-Resistant Steels			
		Materials and Technology, Prague). The Relationship between the Microstructure and the Properties of Some Heat-Resistant Steels and Alloys			
		Materials and Technology, Prague). The Relationship between the Microstructure and the Properties of Some Heat-Resistant Steels and Alloys			
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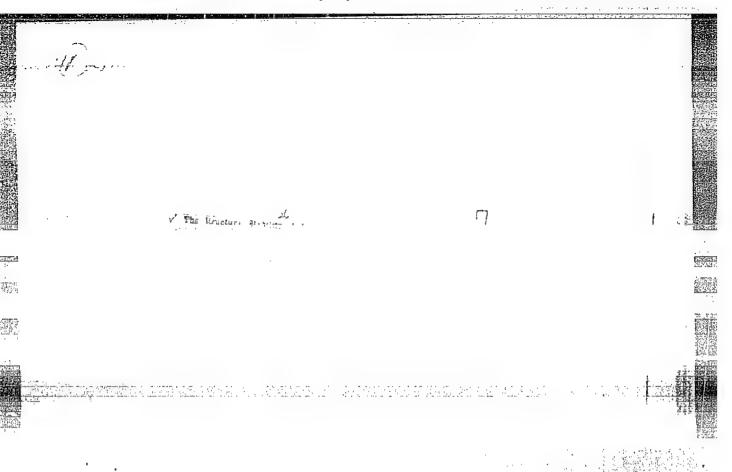
### TUMA, H.

Isolation of carbides from steels and similar alloys. p.909.

CHEMICKE LISTY (Chekoslovenska akademic ved. Cheskaslovensks spotlenost chemicks) Paraha, Czechoslovakis. Vol.53,no.9, Sept.1959.

Monthly List of East European Accessions (EEAI) LC, Vol.9, no.1, Jan.1960.

Uncl.



COUNTRY : Czechoslovakia 8-8 CATEGORY :

ABS. JOUR.; AZKhim., No. 21 1959, No. 74215

7421

AMARCA: Tyklicky, M. and Tume, H.

TITLE : Crystallization of Commercial Fe-Al-C Alloys in

the & -Solid Solution Region

ORIG. PUB.: Hutnicki Listy, 14, No 2, 118-127 (1959)

ABSTRACT : Differential thermal analysis and metallographic

analysis were used in the investigation of the phase diagram of the system Fe-Al-C in the region 15-30% Al.  $\times$  -solid solution was found throughout the temperature range investigated, and in the presence of C the  $\varepsilon$ -phase, graphite,

and Al, C, were also observed.

From authors' summary

CARD: 1/1

E-2 Czechoslovakia COUNTRY CATEGORY 1959, .io. 86228 ABS. JOUR. : AZKhim., so. Dufek, C.; Tuma, H. AUTHOR Rapid Concurrent Potentiometric Determination THST. of Chromium, Vanadium and Iron in Carbides by Means of Trivalent Titanium Chloride. TITLE Hutn. listy, 1959, 14, No 3, 246-247 ORIG. PUB. : ABSTRACT: It was found that on potentiometric titration of a mixture of  $Cr(\ell+)$ , V(5+),  $Mo(\ell+)$  and Fe(3+) with  $TiCl_3$  solution, in a medium of  $HCl_4$  and  $HCl_5$ , the lst jump of potential corresponds to reduction of  $Cr(\ell+)$  to Cr(3+) and Cf(3+) to V(5+) to V(5+) to V(5+)of V(5+) to V(4+), the 2nd -- to conversion of Fe(3+) to Fe(2+), and the 3rd -- to conversion of  $Mo(\ell+)$  to Mo(5+). In the presence of Ra-tartrate or citrate there is observed also a 4th jump of potential corresponding to conversion of V(4+) to V(3+). On analysis of carbides isolated from steel, 10-50 mg sample is dissolved in 3-5 ml (0% EC10m, evaporated by the correct discount of the correct discount d with several drops concentrated HNO3 to evolution of dense fumes (0.5 hour), residue combined with several drops of CARD: 1/2

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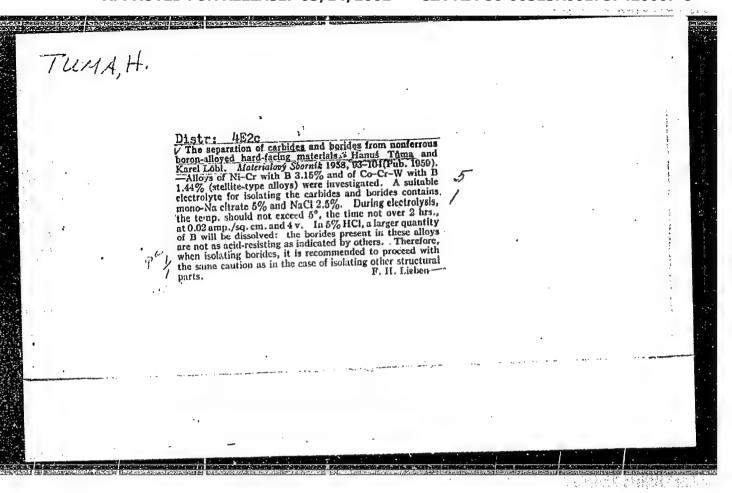
ABS. JOUR. : RZKnim., No. 1959, No. 86228

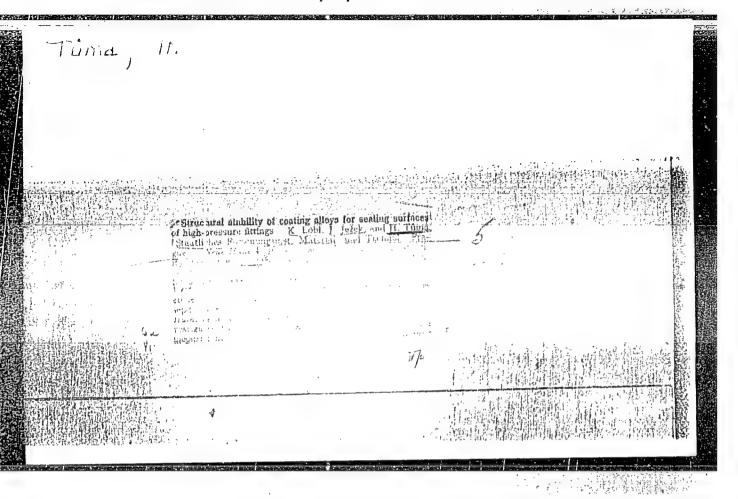
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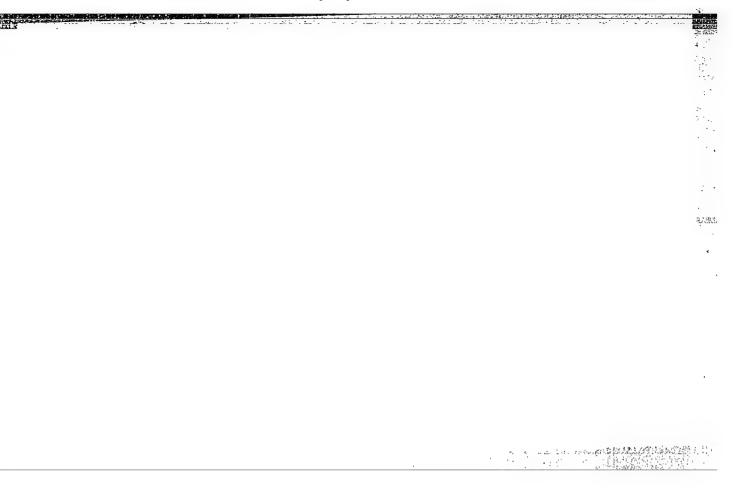
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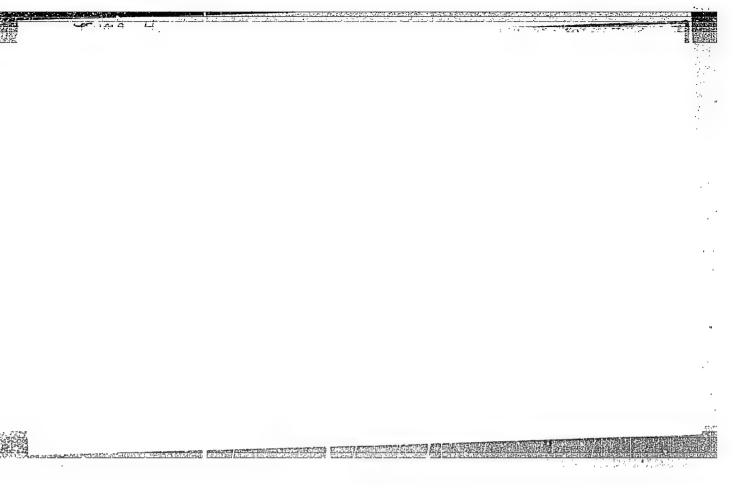
ABSTRACT : water, added 15 ml concentrated HCl, and titrated with with 0.005 N solution TiCl<sub>3</sub>. After titration of Cr(6+), V(5+), Fe(3+) and No(6+), there are added to the same solution 40 ml of 40% solution of Na-citrate and titration is continued to the jump of potential corresponding to V. From the difference between the number of all of TiCl<sub>3</sub> solution used up to get the 1st jump of the potential (Cr+V) and the last jump of potential (V), the content of Cr is determined. -- N. Turkevich.

CARD: 2/2









#### TUMA, H.

Importance of ascorbie acid in analytical chemistry.

p. 390 (Chemie, Vol. 9, no. 3, June 1957, Praha, Czechoslovakia)

Monthly Index of East European Accessions (EEAI) LC. VOL. 7, no. 2, February 1958

TUMA, H.; TOMAN, V.

Methods of evaluating spectro-photometric measurements in a simultaneous determination of two or more components of a solution. p. 621. (MUTHICKE LISTY, Vol. 12, No. 7, July 1957, Brno, Czechoslovakia)

SO: Monthly List of East European Accessions (EEAL) LC, Vol. 6, No. 12, Dec 1957. Uncl.

TUMA, H.

Photocolorimetric determination of zircoium with morin.

p. 722 (CHEMICKE LISTY) Vol. 51, no. 4, Apr. 1957, Praha, Czechoslovakia

SO: Monthly Index of East European Accessions (EEAI) LC, Vol. 7, No. 3, March 1958

Tuma, H.

Titrimetric determination of higher aluminum content in iron alloys. p. 98. HUTNICKE LISTY. (Mihisterstvo hutniho prumyslu a rudnych dolu) Brno. Vol. 11, no. 2, Feb. 1956.

Source: EEAL IC Vol. 5, No. 10 Oct. 1956

TUMA, H.; TIETZ, N.

SCIENCE

Periodical COLLECTION OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS. SECRNIK CHEKHOSLOVATSKIKH KHIMICHESKIKH RABOT. Vol. 23, no. 1, Jan. 1958.

TUMA, H.; TIETZ, N. Photocolorimetric determination of zirconium by means of morin. In German. p. 142.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, no. 3, March, 1959. Uncl.

TUMA, it

CZECHOSLOVAKIA / Chemical Technology. Chemical Prod- H ucts and Their Applications. Corrosion. Corrosion Control.

Abs Jour: Ref Zhur-Khimiya, No 9, 1959, 31574.

Author : Vyklicky, M., Prenosil, B., Tuma, H.

Inst : Not given.

Title : Oxidation of Fe-Al-C Alloys.

Orig Pub: Hutnicke listy, 1958, 13, No 6, 490-496.

Abstract: The results of studying the oxidation of Fe-Al-C

alloys, with a different content of Al and C at a temperature range of 900-1500°, indicated that, after the initial uniform oxidation, some alloys subjected to analysis exhibited an anomalous (A) oxidation. In the A oxidation, blisters were formed on the surface of metals, due to the destruction of the initial protective film of ox-

Card 1/3

CZECHOSLOVAKIA / Chemical Technology. Chemical Prod- H ucts and Their Applications. Corrosion. Corrosion Control.

Abs Jour: Ref Zhur-Khimiya, No 9, 1959, 31674.

Abstract: idation. In places where the blisters took place, oxidation proceeded faster than in the parent metal. It was established that, in the A oxidation, the usual laws pertaining to the development of protective oxidized films do not apply. The duration of the initial uniform oxidation is reduced by a temperature rise, an increased content of C in the alloys and a decreased content of Al. The metallographic investigations of the parent metal under the oxidized film indicated that the destruction of the initially formed oxidized film is connected with volume changes resulting from the transformation of the 2-phase into the 7-phase. However, the transformation

Card 2/3

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

: Tuma, H. Author

: The Separation of Carbides From Highly Alloyed Inst

Title Steels with Bromine.

Orig Pub: Hutnicke Listy, 13, No 8, 717-719 (1958) (in Czech).

Abstract: Optimum conditions for the separation of carbides from highly alloyed steels with bromine solutions have been investigated (10% solutions of bromine nave been investigated (10% solutions of bromine in water, CH3OH, and methylacetate were used). The sample to be analyzed, containing 0.5 gm carbides, sample to be analyzed, containing o.5 gm carbides, is treated with 100 ml of bromine solution, the resulting mixture is kept in a covered vessel at 15-sulting mixture is kept in a covered vessel at 15-180 for 24 hrs or heated to 50°, and filtered through

Card 1/4

79

CZECHOSLOVAKIA / Analytical Chemistry: Analysis of E-2 Inorganic Substances!

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

Abstract: a porous glass filter. The residue is washed with water, ethanol, and ether (when the sample is treated initially with an aqueous bromine solution), methanol and ether (when a methanolic bromine solution is used) or with methylacetate (when a solution of bromine in methylacetate is used), dried at 50° in a vacuum dessicator, and weighed. In the analysis of various carbides (Fe<sub>3</sub>C, Cr<sub>7</sub>C<sub>3</sub>, Mo<sub>2</sub>C, VC, WC, TiC, NbC) it was found that treatment with an aqueous bromine solution at 15° is satisfactory for the separation of Fe<sub>3</sub>C, Cr<sub>7</sub>C<sub>3</sub>, WC, TiC, and NbC; Mo<sub>2</sub>C and VC do not dissolve under these conditions. An increase in the temperature to 50° has a deleterious effect. A methanolic bromine solution has a strong solvating effect on the carbides

Card 2/4

APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001757420007-8"

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

Abstract: than an aqueous bromine solution: Fe<sub>3</sub>C dissolves to the extent of 75% and WC, 12%. In the analysis of various steels containing, in particular, C, Cr, Ni, and Ti, or C, Cr, Ni, Ti, and W it was found that bromine solutions cannot be used to separate carbides from steels containing Mo and V, since the dissolution of the carbides of these metals leads to serious errors. Electrolytic separation methods are recommended for the analysis of Mo- and V-containing steels. Aqueous bromine solutions at 15 can be used in the separation of carbides from steels which do not contain Mo and V. Methanolic bromine solutions are not suited for the analysis of steels containing Fe<sub>3</sub>C, Mo, V, or W and methylacetate bromine solutions cannot be used for the

Card 3/4

80

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

Author : Tuma, H. : Not given. Inst

: The Separation of Carbides From Highly Alloyed Title

Steels with Bromine.

Orig Pub: Hutnicke Listy, 13, No 8, 717-719 (1958) (in Czech).

Abstract: Optimum conditions for the separation of carbides from highly alloyed steels with bromine solutions have been investigated (10% solutions of bromine in water, CH3OH, and methylacetate were used). The sample to be analyzed, containing 0.5 gm carbides, is treated with 100 ml of bromine solution, the resulting mixture is kept in a covered vessel at 15-18° for 24 hrs or heated to 50°, and filtered through

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CIA-RDP86-00513R001757420007-8" APPROVED FOR RELEASE: 03/14/2001

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

Abstract: than an aqueous bromine solution: Fe<sub>3</sub>C dissolves to the extent of 75% and WC, 12%. In the analysis of various steels containing, in particular, C, Cr, Ni, and Ti, or C, Cr, Ni, Ti, and W it was found that bromine solutions cannot be used to separate carbides from steels containing Mo and V, since the dissolution of the carbides of these metals leads to serious errors. Electrolytic separation methods are recommended for the analysis of Mo- and V-containing steels. Aqueous bromine solutions at 15° can be used in the separation of carbides from steels which do not contain Mo and V. Methanolic bromine solutions are not suited for the analysis of steels containing Fe<sub>3</sub>C, Mo, V, or W and methylacetate bromine solutions cannot be used for the

Card 3/4

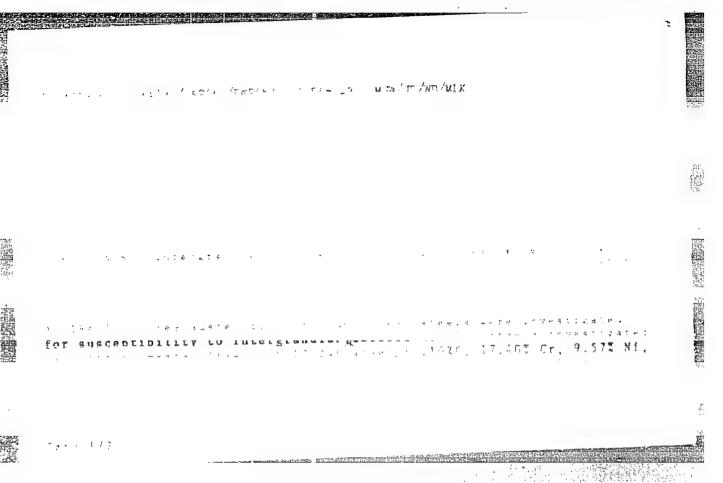
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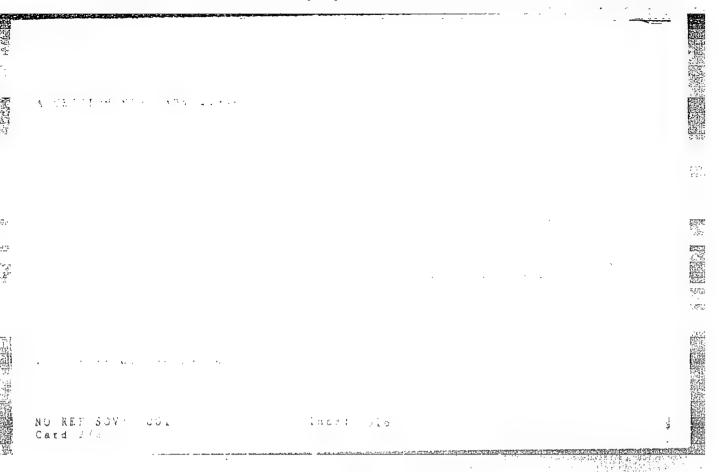
CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2 Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27112.

Abstract: separation of carbides from steels of the type indicated [sic]. -- T. Levi

Card 4/4

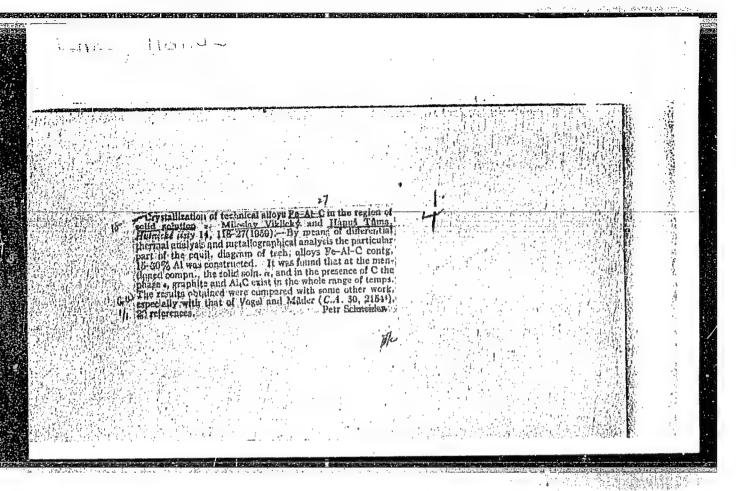


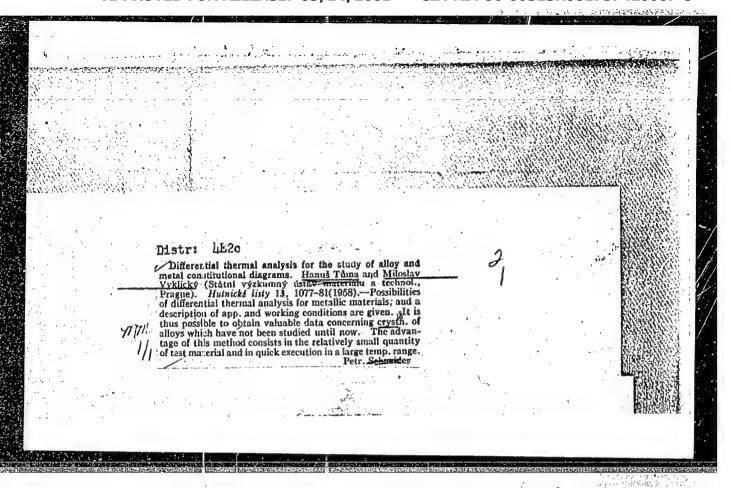


LOBL, Karel; TUMA, Hanus; GROBMER, Pavel

Contribution to the kinetics of carbide segregation in the Cr18Ni9Ti austeritic steels. Hut listy 19 no.12:870-874 D 164.

1. State Research Institute of Material and Technology, Prague.





ACCESSION NR: AP4034555

2/0065/64/000/002/0138/0152

AUTHOR: Lobl, Karel, (Lebl, Karel); Tuma, Hamus (Tuma, Garmsh)

TITLE: Precipitation and solution of carbide M23C6 in type 18/9 low-carbon austenitic steels

SOURCE: Kovove materialy, no. 2, 1964, 138-152

TOPIC TAGS: low-carbon austenitic steel, carbide precipitation, activation energy, nucleation area, diffusion growth, isothermic reasting, plasticity retention, kinetic property, thermodynamic property

ABSTRACT: The advantages of low-carbon steel may be offset by the choice of wrong production methods and by keeping it too long at critical temperatures for the precipitation of carbides. The paper discusses the kinetics of precipitation of chromium carbide in two 40 kg batches of steel of type Crisnia (0.058% and 0.036% C), fairly stable toward phase sigma, when roasted isothermically for a long time at 400—1,000 C, and studies their behavior by direct electrolytic isolation in an alcoholic 5% hydrochloric-acid solution

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ACCESSION NR: AP4034555

and chemical analysis of the carbides. Special attention was paid to the processes in the areas of nucleation (up to 650C) and diffusion growth (650--760/770C) of these carbides and above 760 and 770C, respectively, where carbide M23C6 was partially soluble in the two batches, taken from ordinary production in a basic induction furnace. The activation energy of the carbide reaction was found to be 32,000--38,000 cal/mol in the area of highest solubility; over 70,000 cal/mol in the area of pronounced diffusion growth; around 5,000 cal/mol in the nucleation area. The average chemical composition of M23C6 is Gr19.8 Fe3.2C6 for the state of equlibrium. In the nucleation area the precipitated carbide was substantially richer in iron, whereas in the diffusion-growth area the iron content declined due to the increased rate of diffusion of the chromium. The distribution of total carbon between austenite and carbides depends upon the carbon content above 770C; below this temperature, not only on the carbon content, but also on the time of isothermic roasting. The mechanical experiments with samples subjected

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# ACCESSION NR: N 4034555

to isothermic strain for up to 10,000 hours showed that 18/9-type chromium steels with reduced carbon content retain considerable plasticity even after long roasting. The results of electrolytic isolation of the carbide phases are helpful in studying the kinetic and thermodynamic properties of carbide reactions. Orig. art. has: 10 figures and 4 tables.

ASSOCIATION: Statni vyzkumy ustav materialu a technologie, Prague (State Research Institute for Material and Technology)

SUBMITTED: 12Sep63

DATE ACQ: 11May64

SNCL 2 OO

SUB CODE: MM

NO REF SOV: 004

OTHER: 005

Card 3/3

211111 2/034/61/000/008/001/005 E073/E335

18.1130

Vyklický, Miloslav, Löbl, Karel, Kabrhel, Adolf, Tůma, Hanus, Číhal, Vladimír and Prazák, Milan AUTHORS:

TITLE: Influence of Molybdenum and Copper on the Properties of Stainless Chromium

PERIODICAL: Hutnické listy, 1961, No. 8, pp. 553 - 560

TEXT: According to data published in the literature (Ref. 2 - Copper in Cast Steel and Iron. Copper Development Association, London), high-alloy chromium steels containing 2-3% Si and 1.5-2% Cu have a high resistance to alum and are extensively used in the food-processing industry. An increased C content in chromium steels reduces their resistan to corrosion, particularly after unsuitable heat-treatment. However, low-carbon chromium steels cause difficulties in the manufacture of castings of complex shapes. Therefore, higher C contents are used and the unfavourable influence of the C content is compensated by adding Cu. Although the effect of Mo on chromium steels is known, the authors are not aware of any published information on the combined influence of Cu and Mo Card 1/8

21,111

Z/034/61/000/008/001/005 E073/E335

Influence of Molybdenum ....

on the properties of chromium steels. This is in spite of the fact that such steels are being manufactured, for instance the Czech steel Poldi-AKIBC (chemical composition: 0.12% C, 0.50% Mn, 0.25% Si, 16.15% Cr, 0.20% Mo and 1.75% Cu) and the ferritic chromium steel for use in the chemical industry, containing 0.6-0.8% C, max. 0.7% Mn, max. 2% Si, 28.0 - 30.0% Cr, 2.0 - 2.5% Ni, 2.0 - 2.5% Mo and 2.0% Cu. The authors considered it interesting to investigate the influence of Cu and Mo on the properties of chromium steel and this paper contains the results of these investigations. A total of 11 heats was produced with chemical compositions varying between the following limits: C 0.6 - 0.11%; Cu 0 - 6.11%; Cr 14.58 - 26.6% and Mo 0 - 3.91%. The heats were produced in a 20-kg high-frequency furnace, using as a charge low-carbon steel, low-carbon ferrochromium, low-carbon ferromolybdenum and copper. Of the mechanical properties only the hardness was measured. In agreement with data published in the literature, heats with higher copper contents showed a higher hardness, both

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21,11,1,1

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Influence of Molybdenum ....

in the as-cast and in the annealed states; metallographic tests showed that addition of Cu brought about pronounced structural changes. The corrosion tests were carried out in a number of corrosive media, subdivided into the following groups:

- A. Media with free SO2

  - 1. H<sub>2</sub>SO<sub>3</sub>; 2%; 20 °C 2. NaHSO<sub>3</sub>; 5%; 20 °C
- Organic oxides
  - 3. lactic acid; 10%; 20 °C 4. oxalic acid; 10%; 80 °C 5. citric acid; 10%; 80 °C

  - 80 °c 6. tartaric acid; saturated solution;
  - 7. acetic acid; concentrated; 80

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Influence of Molybdenum ....

- C. Inorganic non-oxiding acids
  - 8. hydrochloric acid; 8%; 20 °C
  - 9. phosphoric acid; 65%; 80 °C
- D. Inorganic Oxiding acids
  - 10. nitric acid; 65%; 80 °C.

A detailed analysis allowed grouping the time dependence of the weight loss due to corrosion into three basic groups: linear dependence (in hydrochloric acid and, in some cases, also in nitric acid at 80 °C); parabolic dependence with steepness increasing with time (NaHSO<sub>2</sub> solution) and, finally,

corrosion rate decreasing with time and characterised by a curve which flattens out. The corrosion tests have shown that steel containing 25% Cr, 2% Mo and 2% Cu had the highest resistance to corrosion, which almost equalled the Czech steel CSN 17241. This type of steel was not investigated in the group of the 17% chromium steels. In the latter steel, Card 4/8

24144

Z/034/61/000/008/001/005 E073/E335

Influence of Molybdenum ....

Mo improved the resistance to corrosion in solutions with free  $\mathrm{SO}_2$ , whilst Cu improved the resistance to corrosion in organic acids. On the basis of laboratory results, SONP Kladno produced two 50-kg heats in a high-frequency furnace with chemical compositions which proved the most favourable in the laboratory tests. The compositions of these heats (in %) were as follows:

Cr Heat Mn Mo Cu A 3829 0.53 0.37 0.019 0.021 15.52 2.05 2.01 0.13 B 3830 0.10 0.54 0.30 0.026 0.017 24.75 1.75

The ingots from both heats were forged into  $250 \times 600 \times 20$  mm blanks and then rolled down to 1 mm thick sheet. These hotrolled sheets were then used in mechanical and corrosion tests and in weldability tests. The most favourable heat-treatment for these steels proved to be the following:



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Influence of Molybdenum ....

Heat A ... 800 °C/0.5 hrs/air

" B ... 900 °C/0.5 hrs/air.

The mechanical properties of thus heat-treated steels do not differ substantially from the properties of semiferritic steels containing 17% Cr (ESN 17041). After this heat-treatment, both heats proved satisfactory in double-bending tests; in Erichsen tests both heats achieved the value of 7.9 mm. Welding tests were carried out by arc-welding in an argon atmosphere; the weldability of Heat A was better than that of Heat B. Potentiostatic polarisation curves were determined to obtain information on the corrosion behaviour of the steels. The following conclusions were reached: Additions of 2% Mo and 2% Cu proved the most suitable. The resistance-to-corrosion of steels with 17% Cr, 2% Mo and 2% Cu is higher than the resistance-to-corrosion of the same type of steel without Mo and Cu. Very good results were obtained with steel containing 25% Cr and an addition of Mo and Cu which, for most corrosive

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2/034/61/000/008/001/005 E073/E335

Influence of Molybdenum ....

media, will have the same resistance-to-corrosion as the austenitic CrNi steel CSN 17241. According to the achieved results, the steel with the lower Cr content can be used for less aggressive corrosion media and in cases in which the steel CSN 17041 cannot be used owing to its lower resistance-to-corrosion or its poor weldability. Steel with a higher Cr content (Heat B) can be used as a substitute for the steel CSN 17241 but the plasticity and weldability of this material are not as good as those of steel CSN 17241. There are 17 figures, 7 tables and 12 references: 6 Czech and 6 non-Czech. The four English-language references quoted are: Ref. 1 - Loring - Metals Handbook, pp. 462 - 465; Ref. 2 - (quoted in text); Ref. 3 - Saklatwalla - Dammler, Trans. Am. Soc. Steel. Treat. 15, 1929; Ref. 4 - Daniloff - The Alloys of Iron and Copper. New York and London, 1934.

X

Card 7/8

2414

Z/034/61/000/008/001/005 E073/E335

Influence of Molybdenum ....

Státní výzkumný ústav materiálu a technologie v

Praze (State Research Institute for Materials

and Technology, Prague)
Státní výzkumný ústav ochrany materiálu
G.V. Akimova v Praze (G.V. Akimov

State Research Institute for the Protection of

Materials, Prague)

SUBMITTED:

ASSOCIATIONS:

November 28, 1960

Card 8/8

CIA-RDP86-00513R001757420007-8" APPROVED FOR RELEASE: 03/14/2001

VYKLICKY, Miloslav; LOBL, Karel; KABRHEL, Adolf; TUMA, Hanus; CIHAL, Vladimir; PRAZAK, Milan

Effect of molybdenum and copper on the properties of chrome stainless steel. Hut listy 16 no.8:553-560 Ag '61.

1. Statni vyzkumny ustav materialu a technologie, Praha (for Vyklicky, Lobl, Kabrhel and Tuma). 2. Statni vyzkumny ustav ochrany materialu G.V. Akimova, Praha (for Cihal and Prazak).

TUMA CZECHOSLOVAKIA/Analycical Chemistry - Analysis of Inorganic HANUS E-2 Substances.

: Ref Zhur - Khimiya, No 8, 1958, 24782 Abs Jour

: Tuma Hanus, Tietz Narcis Author

of Morin.

: Photocolorimetric Determination of Zirconium by Means Inst Title

Chem. 11 ty, 1957, 51, No 4, 722-725; Collect. czechcsl. chem. cc mauns, 1958, 23, No 1, 142-146 Orig Pub

: In an acid medium Zr "forms with morin (I) a soluble lake of lem n-yellow color which is suitable for photometric de-Abstract termination of Zr. Intensity of coloration of the lake in 0.4-0.7 N HCl does not depend on concentration of H 7 but at high temperature and prolonged standing it increases and reaches a maximum after 1 hour. With 0-0.5 mg Zr per

100 ml solution the law of Beer holds. I is used in the form (f an 0.2% solution in CH3OH; 1 ml of this solution

Card 1/2

TUMA, HANUS

CZECHOSLOVAKIA/Optics - Methods of Analysis

K-8

Abs Jour : Ref Zhur - Fizika, No 5, 1958, No 12019

Author

: Toman Vaclay, Tuma Hanus

Inst

: Not Given

Title

: Methods of Processing Spectral Photometric Measurements in Simultaneous Determination of Two or Several Components of

Orig Pub : Hutnicke listy, 1957, 12, No 7, 621-622

Abstract : The author indicates the advantages of simultaneous determination of two or several components (v, Ti, Mo, etc.) in a solution, compared with their separate determination. The method gives economy in chemicals and working time.

Card

: 1/1

CZECHOSLOVAKIA/Optics - Optical Methods of Analysis

K-8

Abs Jour : Ref Zhur - Fizika, No 12, 1958, No 28838

: Turn H., Tietz N. : Not Given Author

Inst

: Photocolorimetric Determination of Zirconium with the Aid of Title

Morin.

Orig Pub : Collect. ezechosl. chem. communs, 1958, 23, No 1, 142-146

Abstract : Translation from Chem. listy, 1957, 51, 722

: 1/1 Card

CIA-RDP86-00513R001757420007-8" APPROVED FOR RELEASE: 03/14/2001

TIETZ, Marcis, inz.; TOMAN, Vaclav, inz.; TUMA, Hamus, inz.

The importance and effect of electrolytes in isolating carbides from steel. Hut listy 12 no.6:517-521 Je '57.

1. Vyzkumny ustav materialu a technologie, Praha.

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্ত্ৰ কৰা মুহুৰ্যুগ্ৰন্থ

CZECH/34-59-8-11/16

AUTHORS: Tuma, Hanus, Engineer and Vyklický, Miloslav, Engineer

TITLE: Isolation of Structural Components in Fe-Al-C Alloys

PERIODICAL: Hutnické listy, 1959, Nr 8, pp 706-710

in a divine la lectro de divini

ABSTRACT: In the here described experiments for developing methods of isolating structural components in Fe-Al-C alloys the

authors studied, in the first instance, the stability of the individual phases in various electrolytes, and the magni-

tudes of the potentials in the respective electrolytes and

they verified the proposed method of isolation. In addition, the problem of isolation of the aluminium carbide Al.C. was solved. On the basis of these and

carbide Al<sub>4</sub>C<sub>3</sub> was solved. On the basis of these and practical tests, the authors propose electrolytic

isolation of the e-phase in Fe-Al-C alloys in a 5% solution

of citric acid with a 2.5% addition of sodium chloride and 0.5 N chlorhydric acid. During the process of

isolation, the temperature should not exceed 3 °C, the current density should not exceed 0.02 Ncm of the

current density should not exceed 0.02 N/cm of the specimen surface and the isolation time should be reduced

Card1/2 to a minimum. A two-hour isolation is practically

Isolation of Structural Components in Fe-Al-C Alloys

adequate for the entire range of the chemical composition of specimens in which the sought structural components (primarily the  $\epsilon$ -phase) occur. For isolating the aluminium carbide  $\mathrm{Al_4C_3}$ , which dissolves in water, an isolation method was evolved which which was based on dissolving the metallic component of the alloy in a waterless solution of bromine in metal acetate. There are 8 figures, 5 tables and 5 references, of which 1 is English and 4 Czech.

ASSOCIATION: Státní výzkumny ústav materiálu a technologie, Praha (State Research Institute for Materials and Technology, Prague)

Card 2/2

TUMA, J.

Soviet nephelometers for the study of aerosols. p. 301.

ZDRAVOTNI TECHNIKA A VZDUCHOTECHNIKA. (Ceskoslovenska akademie ved. Ceskoslovenska vedecka technicka spolecnost pro zdravotni techniku a vzduchotechniku) Praha, Czechoslovakia, Vol. 2, no. 6, 1959.

Monthly List of East European Accession (EEAI), LC Vol. 9, no. 2, Feb. 1960.

Uncl.

34915-66 EWP(t)/ETI

ACC NR: AP6026597

SOURCE CODE: CZ/0034/66/000/002/0129/0131

AUTHOR: Tuma, H.

ORG: SVUM, Prague

TITLE: Calculation of the amount and composition of phases after isolation of carbides from steels with substantial amounts of isolated material

SOURCE: Hutnicke listy, no. 2, 1966, 129-131

TOPIC TAGS: tool steel, bearing steel, carbide phase, metal analysis

ABSTRACT: The author describes a method which is suitable for the determination of the amount and analysis of the basic metal and the carbide phase in phase analyses of steels and alloys. Qualitative isolation of carbides, and their chemical analysis are sufficient for the determination when the content of one of the carbide forming elements in the basic metal material is also known. The method is applicable when the content of carbides is 5-10%; this occurs in tool and bearing steels, and in hard welding alloys. Orig. art. has: 7 formulas and 1 table. [JPRS: 34,779]

SUB CODE: 11 / SURM DATE: none / ORIG REF: 005

Card 1/1 1/195

TUMA, J.

New measurements of two-grade links. p. lil.
SIABORPHOUDY OBZOR. Praha. Vol. 15, no. 1, Jan. 1954.

SOURCE: East European Accessinns List (EEAL), IC, Vol. 5, no. 3, March 1956.

TUMA, J.

Improved Hungarian DR-50 dumper. p. 341.

Vol. 3, no. 10, Oct. 1954 (Mechanisace) INZENYRSKE STAVBY Praha, Czechoslovakia

So: Eastern European Accession Vol.5 No. 4 April 1956

TUMA, J.

Tractor-drawn scrapers. p. 119. Dumpers and bulldozer. p. 123.

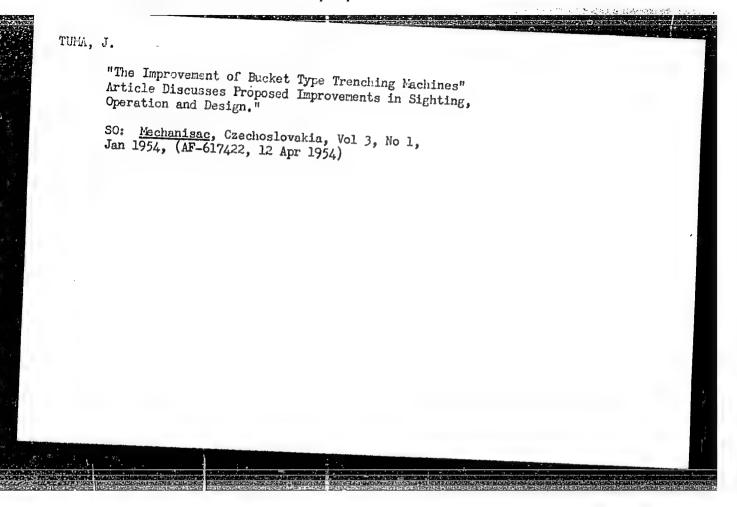
Vol. 3, no. 4, April 1954 (Mechanisace) INZENYRSKE STAVBY Praha, Czechoslovakia

So: Eastern European Accession Vol. 5 No. 4 April 1956

TUMA J. - Mechanisace, Vol. 3, No. 2. Feb. 1954

Improvement of bucket-type trench excavators.p 49

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, No. 9, Sept 1955 Uncl.



TUMA, J.

TUMA, J. Methods of increasing the capacity of machinery for earthwork. p. 544.

Vol. 4, no. 6, 1955 SOVETSKA VEDA: STAVEBNICTVI TECHNOLOGY Praha, Czechoslovakia

So: East European Accessions, Vol. 5, no. 5, May 1956

TUMA, J.

Increasing the efficiency of excavators by using the Soviet experiences. p.343

INZENRYSKE STAVEY. (Ministerstvo stavebnictvi) Praha Vol. 3, no. 8, Aug. 1955

East European Accessions List

Vol. 5 No. 1

Jan. 1956

TUNA, J.

New- Syrovatka-Krchnak hot-air unit. p. 305 (Mechanisace. Preha. Vol. 3, no. 9, Sept. 1954)

SO: Monthly List of East European Accessions, (EEAL), IC, Vol. 4, No. 6, June 1955, Uncl.

TUMA, J.

TUMA, J. Asurvey of the construction and the development of Soviet nuclear reactors. p. 878, Vol. 6, no. 6, 1956 SOVETSKA VEDA: STROJIRENSTVI Praha, Czechoslovakia

SOURCE: EAST EUROPEAN ACCESSIONS LIST (EEAL) VOL 6 NO & APRIL 1957